Vol. 12, No. 2, June 2023, pp. 690~698

ISSN: 2252-8806, DOI: 10.11591/ijphs.v12i2.22742

# Bioaccumulation of heavy metals using aquatic plants in wastewater

## Muliyadi Muliyadi<sup>1</sup>, Purwanto Purwanto<sup>2</sup>, Sri Sumiyati<sup>3</sup>, Mussadun Mussadun<sup>4</sup>

<sup>1</sup>Department of Graduate Program of Environmental Scince, Faculty of School of Postgarduate Studies, Universitas Diponegoro, Semarang, Indonesia

<sup>2</sup>Department of Chemical Engineering, Faculty of Engineering, Universitas Diponegoro, Semarang, Indonesia <sup>3</sup>Department of Environmental Engineering, Faculty of Engineering, Universitas Diponegoro, Semarang, Indonesia <sup>4</sup>Department of Urban and Regional Planning, Faculty of Engineering, Universitas Diponegoro, Semarang, Indonesia

## **Article Info**

## Article history:

Received Nov 19, 2022 Revised Feb 21, 2023 Accepted Mar 10, 2023

## Keywords:

Aquatic plants Bioaccumulation Heavy metals Wastewater

# **ABSTRACT**

Heavy metals are very harmful to the environment if present in excessive concentrations and will have an effect on humans if they interact directly. Knowing the findings of heavy metal bioaccumulation in aquatic plants when reducing pollutants in liquid waste is the main goal of this study. Various types of data were developed and analyzed to answer the specific goals set through the EBSCO, ProQuest, and Science Direct Search Engine, by looking at several parameters. Bioaccumulation data for aquatic plants in liquid waste were obtained from five articles that met the criteria. The findings showed that there are six types of plants that can be used as heavy metal bioaccumulators in wastewater, including Limnobium laevigatum, Salvinia auriculata, Lemna minor, Colocasia esculenta, T. latifolia, and T. palustris. Data was also obtained that most research was carried out using liquid waste originating from within the city with a research time ranging from 1-45 days. The heavy metals that were most commonly accumulated in these plants were Zn, Pb, Cr, Ni, Cu, Fe, Mn, and Cd. It can be concluded that bioaccumulation using aquatic plants is very reliable for various types of heavy metals that can be removed.

This is an open access article under the **CC BY-SA** license.



690

#### Corresponding Author:

Muliyadi Muliyadi

Department of Doctorate Program of Environmental Scince, School of Postgarduate Studies,

Universitas Diponegoro Semarang, Indonesia

Email: muliyadi.blues90@gmail.com; muliyadi@students.undip.ac.id

# 1. INTRODUCTION

Heavy metals are pure metal components found in the Earth's crust and are useful to the Earth itself. However, in many ecosystems, certain heavy metals are toxic to humans and the environment. In recent years, environmental pollution caused by heavy metals has become a problem for all countries, not only for individuals but also for communities in modern society, with very harmful cumulative effects [1], [2]. All human activities produce waste, both internal and liquid waste, especially in the use of water, because almost 80% of water ends up as waste water. There is a lot of waste water management proposed by previous researchers, and if this management is not done properly, it will become a serious problem for the environment in general, and the impact will also be felt by humans. Effective wastewater management is one way to manage liquid waste, so it does not pollute the environment. Waste water treatment has always been an important matter in community practice, this is for reasons of environmental health and the community itself. The importance of knowing the heavy metals in wastewater is not only beneficial for the environment

but is also useful for human survival, where we will be able to determine its consistency, potential changes, make action adjustments if needed, and assess the functionality of the drainage system due to heavy metal contamination.

Heavy metals refer to elements that are metalloids and metals [3]. Various sources of heavy metals in the earth including weathering of rocks, volcanic eruptions, and atmospheric deposits [4]. Heavy metal accumulation can be defined as the melting of heavy metal elements in ecosystems, especially in aquatic ecosystems [5]. The exposure of aquatic plants to toxic levels of heavy metals initiates a wide spectrum of physiological and metabolic changes [6]–[8] which includes decreased plant growth, changes in cell metabolism, damage to cell membranes, protein changes, and inhibition of photosynthesis, leading to senescence and, ultimately, plant death [9], [10]. Many aquatic plants have the ability to tolerate heavy metals in the surrounding water and can even accumulate these metals into their roots and other body parts [11]–[13]. There are also many plants that grow in conditions where the surrounding waters are polluted with heavy metals die due to oxidative stress.

Mitigation technology for heavy metals exposure to the organisms in aquatic environment currently tends to be expensive [14]. In various countries, aquatic plants are often used as bioindicators of water quality [15], [16]. Study from Ali *et al.* [17] demonstrated that aquatic plants accumulate certain metals to varying degrees from one species to another [18]. Many aquatic plants can be used as indicators of water quality [19]. Several studies of aquatic plants have been conducted under laboratory conditions. In a laboratory study of heavy metal phytoremediation by three wetland macrophytes [20], highlighted that all tested species accumulated higher concentrations of the four heavy metals such as Pb, As, Mg, and Hg examined compared to their media. Abdallah [21] investigated two aquatic macrophytes, *Ceratophyllum demersum* and *Lemna gibba*, for the removal of two toxic metals (Pb and Cr) from aqueous solutions.

Many studies have studied heavy metals, both their effects on humans, animals and plants and some of these studies also explain how plants are capable of being bioaccumulators, however, there is no research that thoroughly studies which plants can be used as bioaccumulators of heavy metals in wastewater. Therefore, this meta-analysis will try to find out what types of plants can be used as bioaccumulators in wastewater by looking at some of the supporting parameters such as what heavy metals can accumulate in aquatic plants, how big are the accumulations, how long does it take to accumulate and where are they accumulated.

#### 2. RESEARCH METHOD

The Literature Study used in this writing was 51 articles and a selection was made of the title and parameters to be studied, it was concluded that 5 articles met the requirements to be discussed in depth. The selection process was carried out using the PRISMA technique is a review technique by paying attention to the checklist to be examined such as title, abstract, introduction, method, results, discussion and conclusion. The stages of this research include i) defining the feasibility of features, ii) defining information sources, iii) selecting literature, iv) collecting data, and v) selecting data items. Obtained data can be seen in Figure 1 (see in Appendix). The several databases have been developed to achieve a specific purpose, including peerreviewed reports, original articles, and case studies, and are sourced from journals that have been published in their entirety in academic databases or search engines, such as EBSCO, ProQuest, and Science Direct. In the EBSCO database, four articles matched the search category out of the 15 articles that appeared related to the topic that is wanted to search for with the selected keywords, whereas in the ProQuest database, no related articles were found. The Science Direct database found one out of 36 articles, taking into account the keywords "wastewater, bioaccumulation, and aquatic plants". The data obtained were then combined, reviewed, and analyzed by examining the causation between the influential variables by considering the frequency distribution of the data obtained. Data were also collected and studied in depth to determine the effectiveness of aquatic plants as bio-accumulators and in accumulating heavy metal waste in wastewater by examining several supporting parameters such as plant height and root length. Preliminary data obtained from the search showed that as many as 5 articles were included in the search category, and the data were then sorted based on predetermined parameters such as the ability to accumulate heavy metals, the types of heavy metals that accumulated, the types of aquatic plants that could be used as bio accumulators, the types of waste water used. Contain heavy metals and the long time of accumulation.

# 3. RESULTS AND DISCUSSION

According to several studies Table 1 (see in Appendix) [22]–[26], six types of plants were found that could be used as bioaccumulators of heavy metals in liquid waste, including *Limnobium laevigatum*, *Salvinia auriculata*, *Lemna minor*, *Colocasia esculenta*, *T. latifolia*, and *T. Palustris*. Data was also obtained that most research was carried out using liquid waste originating from within the city with a research time

692 □ ISSN: 2252-8806

ranging from 1-45 days. The heavy metals that were most commonly accumulated in these plants were Zn, Pb, Cr, Ni, Cu, Fe, Mn, and Cd. More details can be found in table.



Figure 2. Types of plants capable of accumulating heavy metals in wastewater Soucre: [22], [23], [24], [25], and [26]

Plants that can be used as bioaccumulators in liquid waste include *Limnobium laevigatum*, *Colocasia esculenta*, *T. latifolia*, *Lemna Minor*, *Salvinia auriculata*, *T. Palustris* can be seen in Figure 2 above. The findings showed that Limnobium laevigatum plants were able to accumulate heavy metals in their roots for the longest time, up to 28 days, with the heavy metals that accumulated being Pb and Cr 1,000 μg, Ni 500 μg, and Zn 100 μg [22]. The longer the exposure time to liquid waste containing heavy metals, the greater the content or accumulation of heavy metals in plants. The heavy metal Cd can also be accumulated by aquatic *Lemna minor* plants with the ability to accumulate 0.332 μg in their roots in just four days. The same thing applies to Pb which can accumulate up to 0.132 μg at the same time [24].

Generally, heavy metals are able to accumulate well by aquatic plants such as *Limnobium laevigatum* [22], *Salvinia auriculata* [23], *Lemna minor* [24], *Colocasia esculenta* [25], *T. latifolia* and *T. palustris* [26]. This is supported by the fact that the use of plants to remove pollutants has been used successfully, by a technique known as phytoremediation [27]–[29]. Some aquatic plants have definitely been used as heavy metal decomposition in the environment as is done in Australia by using Eleocharis dulcis (Chinese water chestnut) to treat uranium mine runoff in Australia [30], and to remove arsenic, previous studies found Neptunia oleracea (water mimosa) to be reliable [31]. Despite the advantages of using aquatic plants to remove heavy metals, they are likely to be harmful to humans by entering the food chain or by direct consumption. Hidden risks that can be caused by heavy metals which are toxic to plants at very low or very high concentrations will cause systemic symptoms [32], [33].

There are various aquatic plants that should be further studied to ascertain their vital role of aquatic plants in accumulating heavy metals. Heavy metal contamination in water bodies that receives less attention is generally located in rural areas and its bioaccumulation into plants which are a source of food for humans is not yet known, one example of an aquatic plant that is often used as food is Nelumbo nucifera (wild lotus). This plant is popular in Chinese food and medicine with all its roots, leaves, fruits, and flowers that can be eaten as medicine [34]. Some aquatic plants tend to accumulate metals biologically depending on their initial concentrations in water. In such cases, the type of metal accumulated may vary depending on the plant species [35]. In some cases, the accumulation of heavy metals in plants can also affect aquaponic systems, a biointegration system in which fish and plants live in a symbiotic environment through a combination of aquaculture and hydroponics [36], [37].

Some heavy metals, such as Fe, Zn, Cu, and Mn, are toxic to humans at higher concentrations, and metals such as mercury, Hg, Cd, As, and Pb are toxic even at the lowest levels [38]. Adverse effects, either directly or indirectly, on human health are numerous. The accumulation of heavy metals in plants is influenced by several factors, such as the plant growth stage, variety of plant species, absorption of elements, degree of tolerance to different contaminants, growth rate, and biomass [35]. Although the accumulation ability of aquatic plants is lower than that of terrestrial plants, they can still accumulate large amounts of metals when they have dense root systems, which increases the surface area for metal uptake [39]. Plant species used as bioaccumulators must have native characteristics, such as fast growth rates, high biomass yields, extensive root systems, tolerance to high salinity and pH, various habitat adaptations, and the ability to accumulate absorbed pollutants [40].

The entry of heavy metals into the environment can come from the natural or artificial environment such as water pollution and soil solution [41]. Aquatic plants are also known as accumulators of heavy metals in the aquatic environment based on several previous studies at several locations [39]. Factors that influence the entry of heavy metals into aquatic plants include the type of plant species, the stage of plant growth, and the types of heavy metals present in the waters themselves [42]. Based on research results, several plants are known to be able to absorb heavy metals, such as *Typhalatifolia plants* [43], water hyacinth (*Eichhornia crassipes*) [44], and *Ceratophyllum demersum* [45] and *Potamogeton malaianus* [46], *Urtica dioica L.* [47]–[49] has the ability to accumulate heavy metals.

The bioaccumulation coefficient is the ratio between the concentration of metals in plant tissues and the surrounding media, in both water and soil. This coefficient varies among plants [50]. Metals that have been absorbed by plant tissues through roots, stems, or leaves can return to the surrounding media. This can occur because of external and internal factors. The absorption of heavy metals by plants is influenced more by biotic and abiotic factors, including temperature, pH, and the ionic population of the water system [51]. The increase in heavy metals in plant parts is due to the binding of heavy metals to plant cells, which creates a gradient across the membrane [52]. With time, as the concentration of metals in the tissue increased, leading to saturation, a subsequent decrease in uptake was effectively observed. This can also occur because of toxic effects caused by metals, such as oxidative stress, which can be caused by Cu, Cd, and Zn [53]. After entering plant roots, metal ions may be stored in roots or transported to shoots, most likely via the xylem, with some suggestion that the phloem may also contribute to this [53].

# 4. CONCLUSION

Interesting findings show that There are six types of plants that can be used as heavy metal bioaccumulators in wastewater: Limnobium laevigatum, Salvinia auriculata, Lemna minor, Colocasia esculenta, T. latifolia, and T. palustris. It was also found that most research is currently being carried out using man-made liquid waste and natural wastewater from waste water treatment plants and also urban sewerage. accumulation in these plants was Zn, Pb, Cr, Ni, Cu, Fe, Mn, and Cd. Analysis of between variables was not carried out because the data obtained varied and it was not possible to get rid of one variable with a variable that was not owned in other studies. For further research, it is expected to take several determining parameters so that homogeneous data is obtained so that it can be tested for differences in each variable.

#### ACKNOWLEDGEMENTS

This article does not get any funding from any party but the authors would like to thank the Graduate Program of Environmental Science, School of Postgraduate Studies, Universitas Diponegoro for their support and encouragement in writing this manuscript.

#### REFERENCES

- P. B. Tchounwou, C. G. Yedjou, A. K. Patlolla, and D. J. Sutton, "Heavy Metal Toxicity and the Environment," 2012, pp. 133– 164. doi: 10.1007/978-3-7643-8340-4\_6.
- [2] M. Nouri and A. E. M. Haddioui, "Assessment of metals contamination and ecological risk in ait Ammar abandoned iron mine soil, Morocco," *Ekológia (Bratislava)*, vol. 35, no. 1, pp. 32–49, Mar. 2016, doi: 10.1515/eko-2016-0003.
- [3] S. J. Hawkes, "What Is a 'Heavy Metal'?," Journal of Chemical Education, vol. 74, no. 11, p. 1374, Nov. 1997, doi: 10.1021/ed074p1374.
- [4] P. C. Nagajyoti, K. D. Lee, and T. V. M. Sreekanth, "Heavy metals, occurrence and toxicity for plants: a review," *Environmental Chemistry Letters*, vol. 8, no. 3, pp. 199–216, Sep. 2010, doi: 10.1007/s10311-010-0297-8.
- [5] A. Alengebawy, S. T. Abdelkhalek, S. R. Qureshi, and M.-Q. Wang, "Heavy Metals and Pesticides Toxicity in Agricultural Soil and Plants: Ecological Risks and Human Health Implications," *Toxics*, vol. 9, no. 3, p. 42, Feb. 2021, doi: 10.3390/toxics9030042.
- [6] M. Jaishankar, T. Tseten, N. Anbalagan, B. B. Mathew, and K. N. Beeregowda, "Toxicity, mechanism and health effects of some heavy metals," *Interdisciplinary Toxicology*, vol. 7, no. 2, pp. 60–72, Jun. 2014, doi: 10.2478/intox-2014-0009.
- [7] F. Ashfaque, A. Inam, S. Sahay, and S. Iqbal, "Influence of Heavy Metal Toxicity on Plant Growth, Metabolism and Its

Alleviation by Phytoremediation - A Promising Technology," *Journal of Agriculture and Ecology Research International*, vol. 6, no. 2, pp. 1–19, Jan. 2016, doi: 10.9734/JAERI/2016/23543.

- [8] A. Pande et al., "Heavy metal toxicity in plants and the potential NO-releasing novel techniques as the impending mitigation alternatives," Frontiers in Plant Science, vol. 13, Sep. 2022, doi: 10.3389/fpls.2022.1019647.
- [9] M. Paunov, L. Koleva, A. Vassilev, J. Vangronsveld, and V. Goltsev, "Effects of Different Metals on Photosynthesis: Cadmium and Zinc Affect Chlorophyll Fluorescence in Durum Wheat," *International Journal of Molecular Sciences*, vol. 19, no. 3, p. 787, Mar. 2018, doi: 10.3390/ijms19030787.
- [10] S. Varma and M. Jangra, "Heavy metals stress and defense strategies in plants: An overview," ~ 608 ~ Journal of Pharmacognosy and Phytochemistry, vol. 10, no. 1, pp. 608–614, 2021.
- [11] P. P. Pant and A. K. Tripathi, "Impact Of Heavy Metals On Morphological And Biochemical Parameters Of Shorea Robusta Plant," Ekologia, vol. 33, no. 2, Jan. 2014, doi: 10.2478/eko-2014-0012.
- [12] S. S. Yadessa Chibssa, "Heavy Metal Concentrations and Physicochemical Characteristics of Effluent along the Discharge Route from Hawassa Textile Factory, Ethiopia," *Journal of Environmental & Analytical Toxicology*, vol. 05, no. 04, 2015, doi: 10.4172/2161-0525.1000285.
- [13] d. A. Al-abbawy, b. M. H. Al-thahaibawi, i. K. A. Al-mayaly, and k. H. Younis, "Assessment of some heavy metals in various aquatic plants of Al-Hawizeh Marsh, southern of Iraq," *Biodiversitas Journal of Biological Diversity*, vol. 22, no. 1, Dec. 2020, doi: 10.13057/biodiv/d220141.
- [14] P. K. Rai, "Heavy Metal Phytoremediation from Aquatic Ecosystems with Special Reference to Macrophytes," Critical Reviews in Environmental Science and Technology, vol. 39, no. 9, pp. 697–753, Aug. 2009, doi: 10.1080/10643380801910058.
- [15] S. Ceschin, V. Zuccarello, and G. Caneva, "Role of macrophyte communities as bioindicators of water quality: Application on the Tiber River basin (Italy)," *Plant Biosystems - An International Journal Dealing with all Aspects of Plant Biology*, vol. 144, no. 3, pp. 528–536, Sep. 2010, doi: 10.1080/11263500903429221.
- [16] S. Ceschin, M. Aleffi, S. Bisceglie, V. Savo, and V. Zuccarello, "Aquatic bryophytes as ecological indicators of the water quality status in the Tiber River basin (Italy)," *Ecological Indicators*, vol. 14, no. 1, pp. 74–81, Mar. 2012, doi: 10.1016/j.ecolind.2011.08.020.
- [17] S. Ali et al., "Application of Floating Aquatic Plants in Phytoremediation of Heavy Metals Polluted Water: A Review," Sustainability, vol. 12, no. 5, p. 1927, Mar. 2020, doi: 10.3390/su12051927.
- [18] P. García F, P. M. J, A. S. OA, and N. T. González, "Phytoremediation of soils contaminated with heavy metals," *Biodiversity International Journal*, vol. 2, no. 4, pp. 362–376, Aug. 2018, doi: 10.15406/bij.2018.02.00088.
- [19] K. Tanwir et al., "Ecophysiology and Stress Responses of Aquatic Macrophytes Under Metal/Metalloid Toxicity," in Plant Ecophysiology and Adaptation under Climate Change: Mechanisms and Perspectives I, Singapore: Springer Singapore, 2020, pp. 485–511. doi: 10.1007/978-981-15-2156-0\_16.
- [20] J. Weiss, M. Hondzo, D. Biesboer, and M. Semmens, "Laboratory Study of Heavy Metal Phytoremediation by Three Wetland Macrophytes," *International Journal of Phytoremediation*, vol. 8, no. 3, pp. 245–259, Sep. 2006, doi: 10.1080/15226510600846798.
- [21] M. A. M. Abdallah, "Phytoremediation of heavy metals from aqueous solutions by two aquatic macrophytes, Ceratophyllum demersum and Lemna gibba L.," *Environmental Technology*, vol. 33, no. 14, pp. 1609–1614, Jul. 2012, doi: 10.1080/09593330.2011.640354.
- [22] D. S. Arán, C. A. Harguinteguy, A. Fernandez-Cirelli, and M. L. Pignata, "Phytoextraction of Pb, Cr, Ni, and Zn using the aquatic plant Limnobium laevigatum and its potential use in the treatment of wastewater," *Environmental Science and Pollution Research*, vol. 24, no. 22, pp. 18295–18308, Aug. 2017, doi: 10.1007/s11356-017-9464-9.
- [23] F. R. Espinoza-Quiñones et al., "Study of the bioaccumulation kinetic of lead by living aquatic macrophyte Salvinia auriculata," Chemical Engineering Journal, vol. 150, no. 2–3, pp. 316–322, Aug. 2009, doi: 10.1016/j.cej.2009.01.004.
- [24] M. A. Khan et al., "Differential Bioaccumulation of Select Heavy Metals from Wastewater by Lemna minor," Bulletin of Environmental Contamination and Toxicology, vol. 105, no. 5, pp. 777–783, Nov. 2020, doi: 10.1007/s00128-020-03016-3.
- [25] T. Bindu, M. M. Sumi, and E. V. Ramasamy, "Decontamination of water polluted by heavy metals with Taro (Colocasia esculenta) cultured in a hydroponic NFT system," *The Environmentalist*, vol. 30, no. 1, pp. 35–44, Mar. 2010, doi: 10.1007/s10669-009-9240-6.
- [26] M. Hejna *et al.*, "Bioaccumulation of heavy metals from wastewater through a Typha latifolia and Thelypteris palustris phytoremediation system," *Chemosphere*, vol. 241, p. 125018, Feb. 2020, doi: 10.1016/j.chemosphere.2019.125018.
- [27] Y. K. Sahu, M. K. Deb, K. S. Patel, P. Martín-Ramos, E. K. Towett, and M. Tarkowska-Kukuryk, "Bioaccumulation of Nutrients and Toxic Elements with Macrophytes," *Journal of Hazardous, Toxic, and Radioactive Waste*, vol. 24, no. 1, Jan. 2020, doi: 10.1061/(ASCE)HZ.2153-5515.0000481.
- [28] H. M. Mustafa and G. Hayder, "Recent studies on applications of aquatic weed plants in phytoremediation of wastewater: A review article," Ain Shams Engineering Journal, vol. 12, no. 1, pp. 355–365, Mar. 2021, doi: 10.1016/j.asej.2020.05.009.
- [29] Z. Wei et al., "Unraveling the Impact of Long-Term Rice Monoculture Practice on Soil Fertility in a Rice-Planting Meadow Soil: A Perspective from Microbial Biomass and Carbon Metabolic Rate," *Microorganisms*, vol. 10, no. 11, p. 2153, Oct. 2022, doi: 10.3390/microorganisms10112153.
- [30] R. A. Overall and D. L. Parry, "The uptake of uranium by Eleocharis dulcis (Chinese water chestnut) in the Ranger Uranium Mine constructed wetland filter," *Environmental Pollution*, vol. 132, no. 2, pp. 307–320, Nov. 2004, doi: 10.1016/j.envpol.2004.04.005.
- [31] N. Atabaki, N. A. Shaharuddin, S. A. Ahmad, R. Nulit, and R. Abiri, "Assessment of Water Mimosa (Neptunia oleracea Lour.) Morphological, Physiological, and Removal Efficiency for Phytoremediation of Arsenic-Polluted Water," *Plants*, vol. 9, no. 11, p. 1500, Nov. 2020, doi: 10.3390/plants9111500.
- [32] Y. Fan, Y. Li, H. Li, and F. Cheng, "Evaluating heavy metal accumulation and potential risks in soil-plant systems applied with magnesium slag-based fertilizer," *Chemosphere*, vol. 197, pp. 382–388, Apr. 2018, doi: 10.1016/j.chemosphere.2018.01.055.
   [33] M. Edelstein and M. Ben-Hur, "Heavy metals and metalloids: Sources, risks and strategies to reduce their accumulation in
- [33] M. Edelstein and M. Ben-Hur, "Heavy metals and metalloids: Sources, risks and strategies to reduce their accumulation in horticultural crops," *Scientia Horticulturae*, vol. 234, pp. 431–444, Apr. 2018, doi: 10.1016/j.scienta.2017.12.039.
- [34] M. M. Uddin, M. C. M. Zakeel, J. S. Zavahir, F. M. M. T. Marikar, and I. Jahan, "Heavy Metal Accumulation in Rice and Aquatic Plants Used as Human Food: A General Review," *Toxics*, vol. 9, no. 12, p. 360, Dec. 2021, doi: 10.3390/toxics9120360.
- [35] F. Duman, Z. Leblebici, and A. Aksoy, "Growth and bioaccumulation characteristics of watercress (Nasturtium officinale R. BR.) exposed to cadmium, cobalt and chromium," *Chemical Speciation & Bioavailability*, vol. 21, no. 4, pp. 257–265, Jan. 2009, doi: 10.3184/095422909X12578511366924.
- [36] A. Graber and R. Junge, "Aquaponic Systems: Nutrient recycling from fish wastewater by vegetable production," *Desalination*, vol. 246, no. 1–3, pp. 147–156, Sep. 2009, doi: 10.1016/j.desal.2008.03.048.
- [37] W. Luo et al., "Effects of cadmium pollution on the safety of rice and fish in a rice-fish coculture system," Environment International, vol. 143, p. 105898, Oct. 2020. doi: 10.1016/j.envipt.2020.105898
- International, vol. 143, p. 105898, Oct. 2020, doi: 10.1016/j.envint.2020.105898.

  [38] J. M. Llobet, G. Falcó, C. Casas, A. Teixidó, and J. L. Domingo, "Concentrations of Arsenic, Cadmium, Mercury, and Lead in

- Common Foods and Estimated Daily Intake by Children, Adolescents, Adults, and Seniors of Catalonia, Spain," *Journal of Agricultural and Food Chemistry*, vol. 51, no. 3, pp. 838–842, Jan. 2003, doi: 10.1021/jf020734q.
- [39] W. Xing, H. Wu, B. Hao, W. Huang, and G. Liu, "Bioaccumulation of Heavy Metals by Submerged Macrophytes: Looking for Hyperaccumulators in Eutrophic Lakes," *Environmental Science & Technology*, vol. 47, no. 9, pp. 4695–4703, May 2013, doi: 10.1021/es303923w.
- [40] J. S. Zavahir and G. Seneviratne, "Role of Microbial Communities for Sustainability," Springer, pp. 235–246, 2021.
- [41] M. A. Ghorab, "Environmental Pollution by Heavy Metals in the Aquatic Ecosystems of Egypt," Open Access Journal of Toxicology, vol. 3, no. 1, Apr. 2018, doi: 10.19080/OAJT.2018.03.555603.
- [42] R. Rucińska-Sobkowiak, "Water relations in plants subjected to heavy metal stresses," Acta Physiologiae Plantarum, vol. 38, no. 11, p. 257, Nov. 2016, doi: 10.1007/s11738-016-2277-5.
- [43] F. Duman, E. Urey, and F. D. Koca, "Temporal variation of heavy metal accumulation and translocation characteristics of narrow-leaved cattail (Typha angustifolia L.)," *Environmental Science and Pollution Research*, vol. 22, no. 22, pp. 17886–17896, Nov. 2015. doi: 10.1007/s11356-015-4979-4.
- [44] S. W. Newete, B. F. N. Erasmus, I. M. Weiersbye, and M. J. Byrne, "Sequestration of precious and pollutant metals in biomass of cultured water hyacinth (Eichhornia crassipes)," *Environmental Science and Pollution Research*, vol. 23, no. 20, pp. 20805–20818, Oct. 2016, doi: 10.1007/s11356-016-7292-y.
- [45] C. S. Romero-Oliva, V. Contardo-Jara, and S. Pflugmacher, "Antioxidative response of the three macrophytes Ceratophyllum demersum, Egeria densa, and Hydrilla verticillata to a time dependent exposure of cell-free crude extracts containing three microcystins from cyanobacterial blooms of Lake Amatitlán, Guatemala," *Aquatic Toxicology*, vol. 163, pp. 130–139, Jun. 2015, doi: 10.1016/j.aquatox.2015.04.001.
- [46] C. A. Harguinteguy, A. F. Cirelli, and M. L. Pignata, "Heavy metal accumulation in leaves of aquatic plant Stuckenia filiformis and its relationship with sediment and water in the Suquía river (Argentina)," *Microchemical Journal*, vol. 114, pp. 111–118, May 2014, doi: 10.1016/j.microc.2013.12.010.
- [47] J. Viktorova et al., "Native Phytoremediation Potential of Urtica dioica for Removal of PCBs and Heavy Metals Can Be Improved by Genetic Manipulations Using Constitutive CaMV 35S Promoter," PLOS ONE, vol. 11, no. 12, p. e0167927, Dec. 2016, doi: 10.1371/journal.pone.0167927.
- [48] N. Ryzhenko, N. I. Yastrebisova, and D. I. Ryzhenko, "Cd and pb in the 'soil-plant' system of holosiyiv green park area in kyiv," Polish Journal of Soil Science, vol. 53, no. 2, pp. 199–210, 2020, doi: 10.17951/pjss/2020.53.2.199.
- [49] V. Dimitrijevic, N. Krstic, M. Stankovic, I. Arsic, and R. Nikolic, "Biometal and heavy metal content in the soil-nettle (Urtica dioica L.): System from different localities in Serbia," Advanced technologies, vol. 5, no. 1, pp. 17–22, 2016, doi: 10.5937/savteh1601017D.
- [50] I. Raskin, R. D. Smith, and D. E. Salt, "Phytoremediation of metals: using plants to remove pollutants from the environment," Current Opinion in Biotechnology, vol. 8, no. 2, pp. 221–226, Apr. 1997, doi: 10.1016/S0958-1669(97)80106-1.
- [51] J. Nyquist and M. Greger, "Uptake of Zn, Cu, and Cd in metal loaded Elodea canadensis," *Environmental and Experimental Botany*, vol. 60, no. 2, pp. 219–226, Jun. 2007, doi: 10.1016/j.envexpbot.2006.10.009.
- [52] M. Greger, "Metal Availability and Bioconcentration in Plants," Heavy Metal Stress in Plants, pp. 1–27, 1999, doi: 10.1007/978-3-662-07745-0\_1.
- [53] S. M. Gallego, M. P. Benavídes, and M. L. Tomaro, "Effect of heavy metal ion excess on sunflower leaves: evidence for involvement of oxidative stress," *Plant Science*, vol. 121, no. 2, pp. 151–159, Dec. 1996, doi: 10.1016/S0168-9452(96)04528-1.

#### **BIOGRAPHIES OF AUTHORS**



Muliyadi is a lecturer at the Environmental Health Department of the Ternate Ministry of Health Polytechnic. He is an expert in wastewater management. He is also active as a writer and researcher and is listed as a postgraduate student at Diponegoro University, Department of Doctor of Environmental Sciences. He can be contacted at email email: Muliyadi.blues90@gmail.com.



Purwanto Description of the Masters Program and Doctoral Program at INPT University, Semarang in 1985. Then he took the Masters Program and Doctoral Program at INPT University of Toulouse, France, in the field of Chemical Engineering so that he succeeded in holding the Degree of Advance Studies (DEA) and Doctoral degrees respectively. -in 1991 and 1994 respectively. The professor who was born in Demak on December 28, 1961 began his career as a lecturer at Diponegoro University in 1985 and is now actively teaching at the Department of Chemical Engineering and the Environmental Science Program at Diponegoro University. He is an expert in Process System Engineering and Chemical Reaction Engineering. For his hard work and perseverance, in 2015 he was appointed Dean of the 2015-2019 Diponegoro University Postgraduate School Program. In addition, he has obtained a position as a Professor from Diponegoro University in 2009 and now serves as Chair of Commission IV of the Academic Senate. He can be contacted at email purwanto.profundip@gmail.com.

696 □ ISSN: 2252-8806



Sri Sumiyati is a lecturer at Diponegoro University with expertise in waste, both solid waste and liquid waste, currently she holds the position of head of the environmental lab at the Faculty of Engineering, Department of Environmental Engineering. He is active in research and has published in many journals, he also teaches in several fields of science, especially in the field of environmental engineering. She can be contacted at email: noesri71@gmail.com.



Mussadun is a lecture at the Department of Urban and Regional Planning, Faculty of Engineering—Diponegoro University. He was graduated from undergraduate program in architecture at the University of Diponegoro in 1996, masters program in coastal and small island resource management at the Bogor Agricultural Institute (IPB) in 2005. Then, he continued his study in IPB and hold the doctoral degree in Management of Coastal Resources and in 2012. Afterwards, he was served as a secretary of the doctoral program in architecture and urban science, Faculty of Engineering, University of Diponegoro. His research concerns in coastal area management and urban design. He can be contacted at email: mussadun@gmail.com.

#### **APPENDIX**

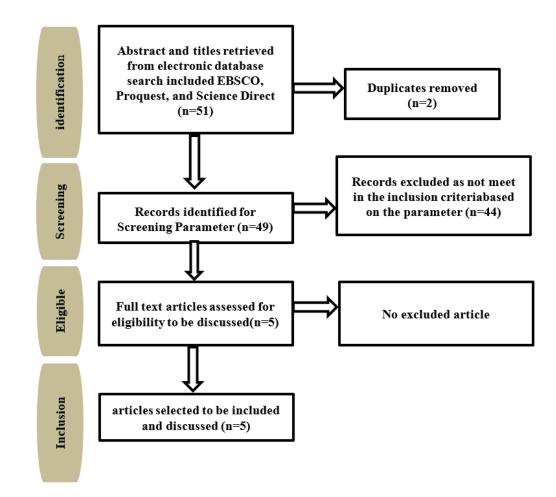


Figure 1. Research flow chart using PRISMA

Table 1. Summary of systematically evaluated studies related to bioaccumulation of heavy metals by using aquatic plants in liquid waste (Continue)

Plant type	Waste source	Accumulation of heavy metals in roots	Accumulation of heavy metals in leaves	Duration	References
Limnobium laevigatum	treatment of wastewater	(T1) were Pb 1 μg Cr 4 μg Ni 25 μg and Zn 30 μg	NM	T1=14 days T2=21 days T3=28 days	[22]
		(T2) were			
		Pb 70 μg Cr 70 μg			
		Ni 70 μg and Zn 70 μg			
		(T3) were			
		Pb 1,000 μg Cr 1,000 μg			
		Ni 500 μg			
Salvinia	Artificial	and Zn 100 µg	(T1) were	T1-1 day	[23]
Salvinia auriculata	wetlands	(T1) were Cr 6.8 μg	Cr 2.1 μg	T1=1 day T2=2 days	[23]
	wastewater	Mn 43μg	Mn 8.6 μg	T3=5 days	
		Zn 74 μg	Zn 17.1 μg	T4=8 days	
		Fe 1,541 μg Cu 49 μg	Fe 176 μg Cu 3.7 μg	T5=15 days T6=22 days	
		· -		T7=29 days	
		(T2) were Cr 5.4 μg	(T2) were Cr 1.4 µg		
		Mn 86 μg	Mn 9.2 μg		
		Zn 114 μg	Zn 11.7 μg		
		Fe 1,591 μg	Fe 64 μg		
		Cu 64 μg	Cu 2.0 µg		
		(T3) were	(T3) were		
		Cr 16.7 μg	Cr 1.7 μg		
		Mn 22 μg Zn 49 μg	Mn 4.5 μg		
		Fe 2,412 μg	Zn 17.8 μg Fe 122 μg		
		Cu 52 μg	Cu 5.5 μg		
		(T4) were	(T4) were		
		Cr 4,2 μg	Cr 3.2 μg		
		Mn 14 μg Zn 33 μg	Mn 5.8 μg		
		Fe 2,159 μg	Zn 16.2 μg		
		Cu 48 μg	Fe 134 μg Cu 6.3 μg		
		(T5) were			
		Cr 4.4 μg	(T5) were		
		Mn 127 μg	Cr 4.0 μg Mn 9.0 μg		
		Zn 74 μg Fe 1746 μg	Zn 16.1 μg		
		Cu 44 μg	Fe 146 μg		
		(T6) were	Cu 6.2 µg		
		Cr 6.0 μg	(T6) were		
		Mn 156 μg	Cr 3.0 µg Mn 19.4 µg		
		Zn 70 μg Fe 1834 μg	Zn 17.6 μg		
		Cu 43 μg	Fe 162 μg		
		· -	Cu 5.7 μg		
		(T7) were Cr 4.7 μg	(T7) were		
		Cr 4. / μg Mn 13 μg	Cr 3.1 μg		
		Zn 25 μg	Mn 5.6 μg		
		Fe 1553 μg	Zn 17.6 μg		
		Cu 31 μg	Fe 181 μg Cu 6.5 μg		
Lemna	Wastewater	Cd 0.332 µg	NM	4 days	[24]
minor		Ni 0.359 μg		<b>y</b>	,
		Pb 0.132 μg			

Table 1. Summary of systematically evaluated studies related to bioaccumulation of heavy metals by using aquatic plants in liquid waste (*Continue*)

using aquatic plants in riquid waste (Commue)								
Plant type	Waste source	Accumulation of heavy metals in roots	Accumulation of heavy metals in leaves	Duration	References			
						Colocasia	Spiked synthetic	NM
esculenta	wastewater		Cd 0.050 µg	-				
T. latifolia	Wastewater	T1	NM	T1=15 days	[26]			
		Zn 105.80 mg/L		T2=45 days				
		Cu 25.92 mg/L		·				
		T2						
		Zn 271.64 mg/L						
		Cu 47.54 mg/L						
Т.	Wastewater	T1	NM		[26]			
palustris	w aste water	Zn 414.67 mg/L	14141		[20]			
		Cu 136.12 mg/L						
		Cu 130.12 Hig/L						
		T2						
		Zn 409.26 mg/L						
		Cu 105.58 mg/L						